

for each solution composition. The average deviation in the groups of four was 0.01%. Single analyses were made on each of the duplicate solutions of calcium sulfate. The average difference between these determinations was 0.009%. The solubility data for sodium and calcium sulfates are given in Tables I and II.

Specific gravity at 25°C. and refractive index at 25°C. were also determined for each saturated solution. It was possible to make these determinations at 25°C. on solutions saturated at higher temperatures because of the negative coefficient of solubility of both salts. None of the solutions saturated at 20°C. showed signs of precipitation during the time

## Letter to Editor

Johnson, Hunt, Neustadt, and Zeleny report that the Stein mill is an efficient dry grinding mill for soybeans, flax, and safflower seeds but not for sunflower seeds. In our laboratory we have been successfully using this instrument for several years in the grinding of sunflower seed samples. Our technique differs in several particulars from that of the above authors. We grind a 50-g. sample for one minute, then transfer the contents to a 5-in. hemispherical, 16-mesh household type of sieve and work the meats through

required to make these physical measurements at 25°C.

### Acknowledgment

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### REFERENCES

1. Bosart, L. W., and Snoddy, A. O., *Ind. Eng. Chem.*, **19**, 506-510 (1927).
2. Seidell, A., "Solubilities of Inorganic and Metal Organic Compounds," Supplement to the Third Edition, p. 482, New York, O. van Nostrand Co. Inc. (1952).
3. Seidell, A., *ibid.*, Third Edition, vol. I, p. 1301 (1940).
4. Hill, A. E., *J. Am. Chem. Soc.*, **59**, 2242 (1937).

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the mesh. This completely breaks up the cake. Hulls remain on top of the wire. If there are only a few whole seeds, they are picked out and discarded. If a substantial amount remains unground, the coarse fraction is reground for 30 seconds and rescreened. Then the two fractions are blended together. By this technique we are able to use a 1-g. sample for the extraction, otherwise following AF 3-54 method for oil content of flax seed.

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## ABSTRACTS . . . . .

R. A. Reiners, Editor

### • Oils and Fats

S. S. Chang, Abstractor  
Sin'itiro Kawamura, Abstractor  
Dorothy M. Rathmann, Abstractor

**The detection of chemical reactions in complex mixtures of food antioxidants by infrared spectrometry.** L. A. Hall (The Griffith Labs., Inc., Chicago, Ill.) and G. L. Clark. *Food Tech.* **10**, 384-6 (1956). Infrared spectrometry has been used as a modern instrumental technique of characterizing complex mixtures of as many as seven constituents in effective food antioxidants with comparison of data for three typical blends with the absorption band frequencies for the individual constituents. It is demonstrated that the infrared spectrum is highly characteristic for each blend and that there is clear evidence of reactions among the constituents such as the formation of lecithin citrate.

**Which antioxidants for your fat-containing foods.** B. N. Stuckey (Res. Div., Eastman Chemical Products, Inc., Kingsport, Tenn.). *Food Eng.* **28**(7), 76, 78 and 198. There are many fat-soluble antioxidants with which the shelf life of animal fats, vegetable oils, dry cereals, essential oils, vitamins, nuts and confections can be extended. Most complaints arising from the use of antioxidants in fats and oils are due to inadequate mixing of the antioxidants into the fat. It is well known that it is not practical to attempt to stabilize a fat that has already started to show substantial peroxide formation.

**The sterol and carbohydrate constituents of the walnut (Juglans regis).** L. Jurd (Western Utilization Res. Branch, Agr. Res. Service, U. S. Dept. of Agr., Pasadena, Calif.). *J. Org. Chem.* **21**, 759-60 (1956). The sterol and carbohydrate constituents of the walnut have been isolated and identified as  $\beta$ -sitosterol,  $\beta$ -sitosteryl-D-glucoside, and sucrose.

**Contributions to the study of marine products. XL. Waxes and triglycerides of sea anemones.** W. Bergmann, S. S. Creighton and W. M. Stokes (Sterling Chem. Lab. and the Bingham Oceanographic Lab., Yale Univ.). *J. Org. Chem.* **21**, 721-28 (1956). Certain lipid fractions from three sea anemones have been investigated. The warm water anemone *Condylactis gigantea* has been shown to contain substantial quantities of

solid lipides. These were found to consist of a mixture of myristyl myristate and myristyl palmitate and symmetrical palmityl dimyristin. Two cold-water anemones *Bolocera iuediae* and *Actinostola collosa* have been shown to contain substantial quantities of liquid lipides. These were found to consist mainly of esters of unsaturated alcohols and acids of the order  $C_{20}$  and  $C_{22}$  and triglycerides of acids of the same order. Two new alcohols, 11-eicosenol and 11-docosenol, have been isolated. Cholesterol has been shown to be the principal sterol of the cold-water anemones.

**An improved procedure for the isolation of neoabietic acid from pine oleoresin and rosin.** Virginia M. Loeblich and R. V. Lawrence (Naval Stores Res. Sec., U. S. Dept. of Agr., Olustee, Florida). *J. Org. Chem.* **21**, 610-11 (1956). A method is described for isolating neoabietic acid from pine oleoresin or rosin in 8-12% yield by recrystallization of the 2-amino-2-methyl-1,3-propanediol salt of the resin acids from methyl ethyl ketone.

**A note on component fatty acids of the oil from the seeds of *Momordica charantia*, Linn.** J. P. Verma and J. S. Aggarwal (Natl. Chem. Lab. of India, Poona). *J. Indian Chem. Soc.* **33**, 357-8 (1956). The fatty acid composition of the oils from the seeds of *Momordica charantia*, Linn. as obtained by the present analysis is  $\alpha$ -elaeostearic acid, 46.7%; linoleic acid, 7.7%; oleic acid, 15.8%; and stearic acid, 29.8%.

**Fat absorption studies. IV. Polyunsaturated fatty acids in the feces of infants.** L. Söderhjelm. *Acta Soc. Med. Upsalensis* **57**, 438-47 (1952). In the feces of breast-fed infants 1.8-7.1% of the fatty acids contained more than one double bond, compared with 0.4-2.0% in the feces of infants fed with cow milk; the value for meconium was about 3%.

**V. Polyunsaturated fatty acids in fetal organs.** *Ibid.* 448-54. Polyunsaturated fatty acids were estimated in brain, heart, muscle, liver, lung, kidney, and subcutaneous fatty tissue of fetuses obtained from surgical abortions and premature infants. In all the organs examined appreciable amounts of tetra- and hexaenoic acids were always present, but dienoic acids were not always found and occurred rarely in brain tissue. (*C. A.* **50**, 7973)

**The chemistry of branch-chain fatty acids. II.** K. E. Schulte, W. Weisskopf, and J. Kirschner (Deut. Forsch. Lebensmitt,